# **O-Level Chemistry**

# Paper 4

**Unsolved Topical** 

Past Papers With Marking Scheme
According to New Syllabus (2023-2025)

**2014-2021** 

All rights reserved. No part of this publication may be reproduced, Stored in a retrieval system or transmitted, in any form or by any means, electronic, mechanical, photocopying, recording or otherwise, without the prior written permission of the publisher.

Title O-LEVEL TOPICAL CHEMISTRY P4

**Published by** MS Books (042-35774780)

**Legal Advisor** Ashir Najeeb Khan (Advocate)

**AKBAR LAW CHAMBERS** 

39-40, 1<sup>st</sup> Floor, Sadiq Plaza, The Mall, Lahore.

0307-4299886, 042-36314839

For Complaints/Order MS Books

83-B Ghalib Market Gulberg III Lahore

info@msbooks.pk

(042-35774780), (03334504507), (03334548651)

### **PREFACE**

Excellence in learning cannot be claimed without application of concepts in a dexterous way. In this regard one of the logical approach is to start in chunks; like chapter wise learning and applying the concept on exam based questions.

This booklet provides an opportunity to candidates to practice topic wise questions from previous years to the latest. Extensive working of Team MS Books has tried to take this booklet to perfection by collaborating with top of the line teachers.

We have added answer key / marks scheme at the end of each topic for the candidate to compare the his/her answer to the best.

MS Books strives to maintain actual spacing between consecutive questions and within options as per CAIE format which gives students a more realistic feel of attempting question.

Review, feedback and contribution in this booklet by various competent teachers of a subject belonging to renowned school chains make it most valuable resource and tool for both teachers and students.

With all belief in strength of this resource material I can confidently claim that it is worth in achieving brilliance.

Our sincere thanks and gratification to **Mr. Kamal Ahmad** who took out special time to help compile and manage this booklet. We would also like to appreciate chemistry faculty for reviewing and indorsing it.

#### **REVIEWED & RECOMMENDED BY**

#### Muhammad Ali

University of Cambridge, University of Wales LGS Defence (Phase 1&5), LGS Paragon, LACAS, BCCG, The City School CLC 0321-8859967

#### Waqar Ahmad

LACAS, ROOTS Millenieum, Frobels International 0334-9543124

#### Kashif Ali Sehgal

LGS DHA & JT, City RAVI, ROOTS Millenieum, EX-BSS, EX-Aitchison 0322-4828628

#### **Tanvir Gill**

BDC & BCCG 0301-4574832

#### Zafar Igbal

LACAS (Barki & JT) BSS ALJT, KIMS, Crescent, LGS Paragon, ROOTS IVY DHA Ph.5 0333-4227604

#### **Qammar Fayyaz**

LGS JT (Boys & Girls), City RAVI 0300-4266857

#### Kamal Ahmad

LGS, BSS, CBS 0333-4567757

### PAPER P4

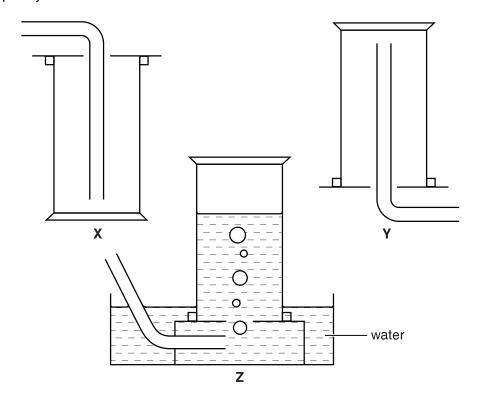
## **CONTENT TABLE**

Ch#	Topics	Pg#
1.	1.1 Experimental Design	7
	1.2 Methods of purification and analysis	25
	1.3 Identification of lons and Gases	46
2.	Particulate Nature of Matter	108
	2.1 Kinetic Particle Theory	112
	2.3 Structure and Properties of Materials	114
3.	Formulae, Stoichiometry and the Mole Concept	116
4.	Electrolysis	257
5.	Energy from Chemicals	281
6.	Chemical Reaction	310
_	6.1 Rate of Reaction	321
	6.2 Redox Reaction	380
	6.3 Reversible Reactions	382
7.	The Chemistry and uses of Acids, Bases and Salts	384
	7.2 Preparation of Salts	393
8.	The Periodic Table	405
9.	Metals	408
10.	Atmosphere and Environment	425
11.	Organic Chemistry	
	11.1 Alkanes	427
	11.2 Alkenes	432
	11.3 Alcohols	440
	11.4 Carboxylic acids	446
12.	Multi-Topic/Challenging Questions	469

### **Experimental Design**

Q2(c)/41/M/J/14

1 (c) Suggest which method, **X**, **Y** or **Z**, is most suitable for the collection of each of the gases NH<sub>3</sub> and HC*l*. Explain your answers.



NH <sub>3</sub>	
HC1	
explanation	
	[4

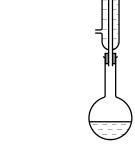
Q5/42/O/N/14

2 Which apparatus is used to separate a mixture of ethanol and water?

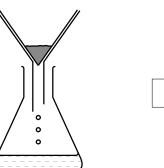
(a)

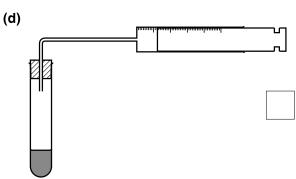
(b)

(1)



(c)





[Total: 1]

Q3/41/M/J/15

3 With which piece of apparatus should a safety bulb be used?

- (a) beaker
- (b) burette
- (c) measuring cylinder
- (d) pipette



[Total: 1]

Q3/42/M/J/15

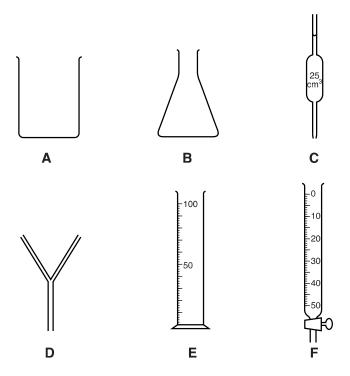
4 Which apparatus should a student use to measure 25.0 cm<sup>3</sup> of a liquid?

- (a) beaker
- (b) conical flask
- (c) measuring cylinder
- (d) pipette

[Total: 1]

Q1/41/M/J/16

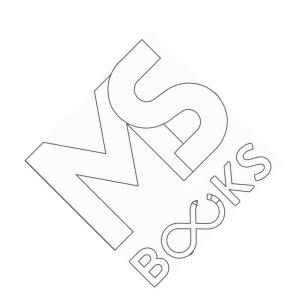
5 (a) The following apparatus is found in a laboratory.



Write in the table the letter of the apparatus most suitable for the purpose.

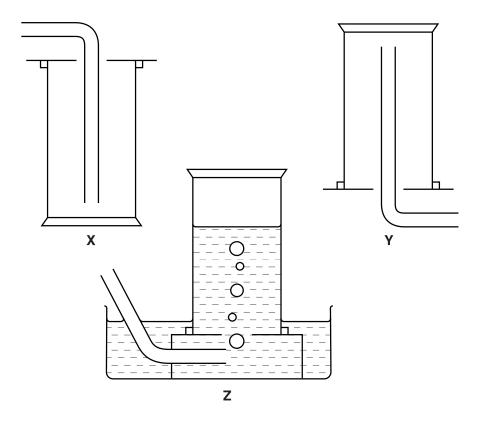
purpose	apparatus
removing 25.0 cm <sup>3</sup> of a liquid from a container	
measuring 60 cm <sup>3</sup> of a liquid	
as a titrating flask	
separating a precipitate from a solution	

[4]



[Total: 7]

(b) The diagrams show three methods for collecting gases.



Which method X, Y or Z is suitable for collecting a gas which is

(i) less dense than air and soluble in water,

	[41]	
***************************************	[1]	

(ii) more dense than air and soluble in water,

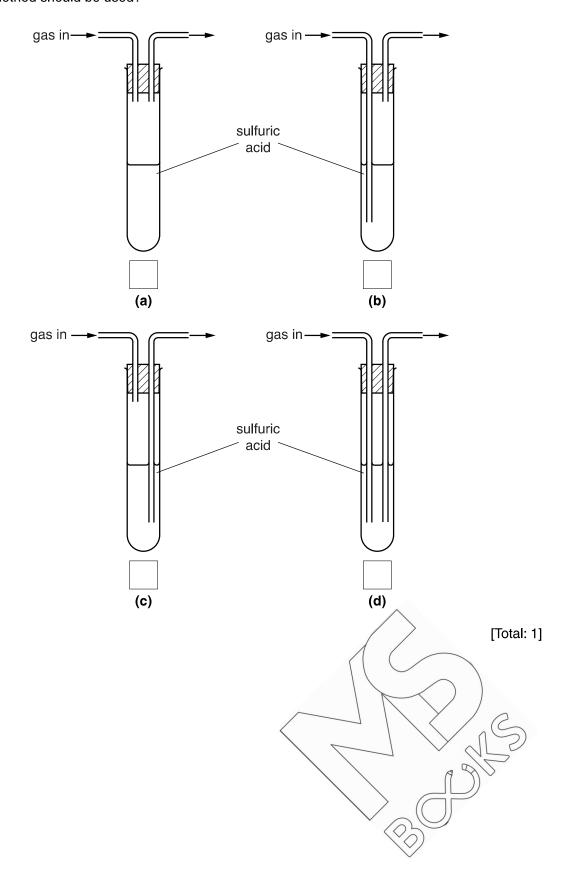
(iii) insoluble in water?



MS Books (O/A Level Notes & Past Papers) www.msbooks.pk (04235774780, 03334504507, 03334548651)
10

Q3/41/M/J/16

**6** A gas may be dried by passing it through concentrated sulfuric acid. Which method should be used?



Q2(c)/41/M/J/14

1 **(c)**  $\mathbf{Y}$  (NH<sub>3</sub>) (1);  $\mathbf{X}$  (HCl) (1)

Both soluble in water (1)

HCl is more dense than air AND NH<sub>3</sub> is less dense than air (1)

[4]

Q5/42/O/N/14

**2** (a) (1)

[Total: 1]

Q3/41/M/J/15

3 (d)

[Total: 1]

Q3/42/M/J/15

**4 (d)** (1)

[Total: 1]

Q1/41/M/J/16

Question 5	Answer	Marks
1(a)	C (1) E (1) B (1) D (1)	4
1(b)(i)	Y	1
1(b)(ii)	X	1
1(b)(iii)	z	1

Q3/41/M/J/16

Question 6	Answer	Marks
3	В	1

